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AN ISOMER OF XANTHANOL FROM *XANTHIUM ORIENTALE**

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Key Word Index—*Xanthium orientale*; Compositae; Heliantheae; sesquiterpene lactones; xanthanolides.Abstract—The aerial parts of *Xanthium orientale* afforded an isomer of xanthanol.

From the genus *Xanthium* several xanthanolides were isolated [1–11], which seem to be characteristic for this genus, though this type of sesquiterpene lactone has been reported from other genera too. We now have re-investigated the aerial parts of *Xanthium orientale* from which the presence of xanthine was reported [7]. Again, only xanthanolides were isolated, the ketones 1 [3] and 2 [2] and the hydroxy acetate 5, which was not identical with xanthanol (3) [1]. The ^1H NMR data of 3 and of the corresponding diacetate 4 were nearly identical with those of the new lactone and the diacetate (Table 1), but there were small characteristic differences in the couplings of H-2 and H-3. Also some of the chemical shifts were slightly different. By spin decoupling, all signals could be assigned. As the couplings of the ring protons were identical with those of 3 and 4, the only possible difference was in the stereochemistry at C-2 or C-4, respectively. The observed shift differences of H-5, however, would favour epimers at C-2. Different stereochemistry at C-10 was unlikely as the couplings of H-10 and the chemical shift of H-14 were nearly identical in 4 and 6. Also a C-8 epimer was not possible, as these epimers characteristically differ in the ^1H NMR spectra [12]. Therefore the new lactone most probably is 2-epixanthanol (5). Unfortunately, the configuration at C-2 is not known for xanthanol and related lactones, which, however, all seem to have the stereochemistry of xanthanol at C-2, if $J_{2,3}$ is conclusive. These results again show that the chemistry of the genus *Xanthium* is very uniform. This genus is placed in the

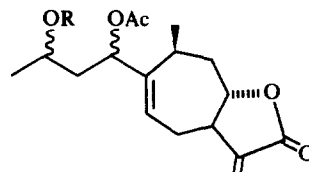
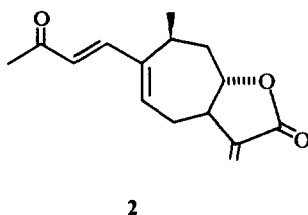
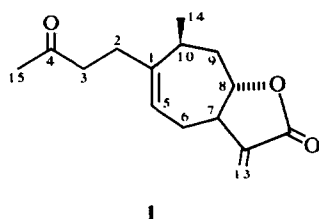
Heliantheae, subtribe Ambrosiinae, a very distinctive group [13], mainly characterized by pseudoguaianolides;

Table 1. ^1H NMR spectral data of compounds 4, 5 and 6 (400 MHz, CDCl_3 , TMS as internal standard)

	4	5	6
H-2	5.22 <i>dd</i>	5.41 <i>dd</i>	5.19 <i>dd</i>
H-3	2.10 <i>m</i>	1.85 <i>ddd</i>	1.88 <i>ddd</i>
H-3'	1.72 <i>m</i>	1.59 <i>ddd</i>	1.75 <i>m</i>
H-4	4.85 <i>ddq</i>	3.75 <i>ddq</i>	4.91 <i>ddq</i>
H-5	5.87 <i>br dd</i>	5.96 <i>br dd</i>	5.93 <i>br dd</i>
H-6 α	2.53 <i>ddd</i>	2.57 <i>ddd</i>	2.53 <i>ddd</i>
H-6 β	2.10 <i>m</i>	2.14 <i>ddd</i>	2.13 <i>ddd</i>
H-7	2.52 <i>m</i>	2.48 <i>ddd</i>	2.44 <i>ddd</i>
H-8	4.38 <i>ddd</i>	4.52 <i>ddd</i>	4.28 <i>ddd</i>
H-9 α	1.72 <i>m</i>	1.76 <i>ddd</i>	1.79 <i>m</i>
H-9 β	2.33 <i>ddd</i>	2.35 <i>ddd</i>	2.31 <i>ddd</i>
H-10	2.79 <i>ddq</i>	2.82 <i>ddq</i>	2.79 <i>ddq</i>
H-13	6.17 <i>d</i>	6.20 <i>d</i>	6.17 <i>d</i>
H-13'	5.45 <i>d</i>	5.48 <i>d</i>	5.44 <i>d</i>
H-14	1.11 <i>d</i>	1.18 <i>d</i>	1.10 <i>d</i>
H-15	1.25 <i>d</i>	1.23 <i>d</i>	1.26 <i>d</i>
OAc	2.06 <i>s</i> 2.04 <i>s</i>	2.12 <i>s</i>	2.04 <i>s</i> 2.03 <i>s</i>

*Part 365 in the series "Naturally Occurring Terpene Derivatives". For Part 364 see Dominguez, X. A., Franco, R., Cano, G., Bapuji, M. and Bohlmann, F. (1981) *Phytochemistry* **20**, 2297.

J (Hz): 2,3 = 3; 2,3' = 10.5; 3,3' = 13.5; 3,4 = 10; 3',4 = 3; 4,15 = 6.5; 5,6 α = 9.5; 5,6 β = 3; 6 α ,7 = 2; 6 β ,7 = 10; 7,8 = 10; 7,13 = 3; 8,9 α = 12; 8,9 β = 3; 9 α ,9 β = 12.5; 9 α ,10 = 9 β ,10 = 3.5; 10,14 = 7; (compound 4: 2,3 = 2,3' = 7).



3 R = H 5 R = H(2-*epi*)
4 R = Ac 6 R = Ac(2-*epi*)

nevertheless, xanthanolides are present in *Iva* species [14, 15] also placed in this subtribe.

EXPERIMENTAL

The fresh aerial parts (120 g) (Botanical Garden Berlin-Dahlem) were extrd with Et₂O-petrol, 1:2, and the resulting extract was sepd by CC (Si gel). The fractions obtained with Et₂O were further sepd by TLC (Si gel, Et₂O-petrol, 3:1, ×2) affording 5 mg 1, 10 mg 2 and 2 mg 5.

2-Epixanthanol (5). Colourless gum, MS *m/z* (rel. int.): 248.140 (M - AcOH, 5) (C₁₅H₂₀O₃), 230 (248 - H₂O, 7), 204 (248 - C₂H₄O, 100), 189 (204 - Me, 33), 176 (204 - CO, 43); CI (isobutane): 309 (M + 1, 72), 249 (309 - AcOH, 100), 234 (249 - Me, 87), 205 (249 - C₂H₄O, 86).

To 2 mg 5 in 1 ml CHCl₃ were added 5 mg 4-pyrrolidinopyridine [16] and 0.1 ml Ac₂O. Usual work-up after 12 hr and TLC (Et₂O-petrol, 1:1) afforded 1.5 mg 6, colourless gum, IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 1785 (γ -lactone), 1750, 1250 (OAc); MS *m/z* (rel. int.): 308 (M - ketene, 2), 293 (308 - Me, 22), 290.152 (M - AcOH, 18) (C₁₇H₂₂O₄), 248 (290 - C₂H₄O, 42), 230 (290 - AcOH, 51), 215 (230 - Me, 14), 175 (100).

$$[\alpha]_{24}^{\text{D}} = \frac{589}{-43} \frac{578}{-44} \frac{546}{-50} \frac{436 \text{ nm}}{-81} \quad (c = 0.07, \text{CHCl}_3).$$

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